# Tutorial: Molecular structures and MD of small and larger molecules using DFTB+

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How to connect to Lired:

- 1. ssh sirle@lired.iacs.stonybrook.edu
- 2. Password: DFTBwork
- 3. You should see prompt "[sirle@lired ~]\$
- 4. Create your own directory, e.g. "mkdir joe" where joe is the directory you will run calculations in
- 5. cd "joe"

How to visualize .xyz data files:

- a) MOLDEN
- b) JMOL
- c) ...

Short instructions for important UNIX commands (I):

#### **General UNIX commands:**

cd <dir></dir>	change directory to <dir></dir>	
cd	Go back one directory	
ls	list files in current directory	
vi <file></file>	view and edit file <file> (gedit is also available)</file>	
cp <file> <new< td=""><td>v&gt; copy file <file> to file <new></new></file></td></new<></file>	v> copy file <file> to file <new></new></file>	
mkdir <dir></dir>	create directory <dir></dir>	

(ask your friends for help if you are lost at any time)

Short instructions for important UNIX commands (II):

#### Important commands:

dftb+(the DFTB+ program)waveplot(plotting molecular orbitals, electron densities, etc.)gnuplot(the gnuplot program)

Visualization: DFTB+ produces .xyz files which can be viewed with MOLDEN or JMOL or ...

To install MOLDEN on your computers, open: http://qc.chem.nagoya-u.ac.jp

Click on "Teaching"

#### You will see:

1. Lecture I – Introduction (PDF)

2.1 Lecture II – Use your computer, see chemistry (PDF)

2.2 Example outputs: h2o.out h2o-freq.out benzene.out nitrobenzene.out sulflower.out sulflower-LMO.out

2.3 Molden program for MacOSX: molden5.4\_MacOSX.tar.gz, for Windows:

molden5.0.7\_native\_windows.zip or molden\_windows\_nt\_95.zip

2.4 Assignment 1 (PDF)

2.5 Introduction to UNIX (Jananese)

To download, you are asked to provide a userid and password.

Userid: qcguest Password: qcigf! (quantum chemistry is great fun!)

# For visualization of molecular orbitals or densities, you can use VMD or any other software that can plot Gaussian 's "cube" files



VMD is a molecular visualization program for displaying, animating, and analyzing large biomolecular systems using 3-D graphics and built-in scripting. VMD supports computers running MacOS X, Unix, or Windows, is distributed free of charge, and includes source code. (more details...)

http://www.ks.uiuc.edu/Research/vmd/

#### Directories on the server

The following directory structure exists with corresponding inputs for DFTB+ in our home directory /gpfs/home/sirle:

Directory	Description
H2O	Water molecule geometry optimization
formamide	input for formamide rotational barrier calculation
benzene	benzene molecular geometry optimization
C60	C <sub>60</sub> geometry optimization
nanotube	(5,5) nanotube geometry optimization
Fe38	Fe38 geometry optimization
C82	single point calculation for MO visualization
Sc2C2C82	single point calculation for MO visualization
MD	water NVE/NVT BOMD simulations

## **Frequently occurring problems**

The location of the DFTB parameter files has to be specified in the dftb\_in.hsd input file. If the specified directory does not exist or the parameter files do not exist, the program will stop with an error message. The parameter files are stored in the directory

#### /gpfs/home/sirle/SKfiles

The default maximum number of geometry updates during geometry optimization is 200. You may have to change this to converge your geometries by setting MaxSteps = 300 (a number larger than 200) in the Driver{} Section of dftb\_in.hsd

# 1a) H2O single point calculation (input)



```
dftb_in.hsd: Geometry = GenFormat {
             <<< "h2o.gen"
             }
             Driver = \{\}
             Hamiltonian = DFTB{
               Charge = 0
               SpinPolarisation{}
               SCC = Yes
               SCCTolerance = 1e-6
               MaxSCCIterations = 100
               Mixer = Broyden{}
               MaxAngularMomentum = {
                O = "p"
                H = "s"
             }
             SlaterKosterFiles = Type2FileNames {
               Prefix = "/gpfs/home/sirle/SKfiles/"
               Separator = "-"
               Suffix = ".skf"
              LowerCaseTypeName = No
             }
             }
                                                 9
```

## 1a) H2O single point calculation (output)

	detailed.out:	Net atc	omic charges (e)
		Atom	Net charge
		1	-0.56908879
_		2	0.28454439
		3	0.28454439
		Eigenv	alues /eV
dftbp.out:		-23.02	2217599
•	te ale ale ale ale ale ale ale ale ale al	-10.69	768502
***************************************	* * * * * * * * * * * * * *	-9.04	017641
**************************************	****	-7.06	500997
		7.809	946767
iSCC Total electronic Diff electronic SCC error		13.40	675941
1 -0.41146057E+01 0.0000000E+00 0.77431585	E+00		
2 -0.41189864E+01 -0.43807053E-02 0.56529143	E+00	Fillings	5
3 -0.41241083E+01 -0.51219356E-02 0.62694765	E-02	2.00	000
4 -0.41241090E+01 -0.64734819E-06 0.72216815	E-04	2.00	000
5 -0.41241090E+01 -0.85880636E-10 0.52564193	E-08	2.00	000
>> Charges saved for restart in charges.bin		2.00	000
		0.00	000
Total Energy: -4.0698274647 H		0.00	000
i otal iviermin free energy: -4.06982/464/ H		0.00	

## 1b) H2O geometry optimization





#### 1c) rotational barrier in formamide

#### 1d-1) water dimer geometry optimization



Task: Make appropriate input by yourself using H2O optimization (task 1b).

Compute SCC-DFTB binding energy  $BE = -[E_{dimer} - 2*E_{monomer}]$ For this conformation: Edimer = -8.1611635663 ha BE = -[-8.1611635663 - 2\*(-4.0779379339)] ha = .0052876985 = 3.3 kcal/mol

## Slater-Kirkwood dispersion term for SCC-DFTB\*

\*Elstner, M. et al., J. Chem. Phys., 114, 5149-5155, (2001).

$$E_{dis}(R_{\alpha\beta}) = -\sum_{\alpha\beta} f(R_{\alpha\beta})C_{6}^{\alpha\beta}(R_{\alpha\beta})^{-6}$$

 $R_0$ : cufoff distance  $N_a$ : effective number of electrons

PO<sub>4</sub>), and S (not SO<sub>2</sub>)

fRI/(R\*\*6) -Damping function 4.5 R Available for C, H, N, O, P (only Pauli repulsion ~ 1/R<sup>6</sup> behavior

#### benzene/inputs> ls benzene-disp benzene-nodisp PD-disp S-disp T-disp T-shape T Sandwich S Parallel-displaced PD

Task 1:

optimize benzene monomer with and without dispersion interaction; what differences do you observe?

Task 2: Using dispersion interaction, optimize S, PD, and T dimer structures and compare your energies with literature values:

	S	Т	PD		
BE	1.65	2.69	2.67	[kcal/mol]	
Note that DFTB-D interactions are in the range 3-5 kcal/mol.					

2a) C<sub>60</sub> geometry optimization

# Directory: C60 From T<sub>n</sub>=2000K MD simulation DFTB optimization **Initial geometry:** -103.1220165514 H Total Energy:

Total Mermin free energy: -103.1220165514 H Maximal force component: 0.639238E-01

#### Optimized geometry, 18 geom. updates:

Total Energy: -103.2321663552 H Total Mermin free energy: -103.2321663552 H Maximal force component:  $0.693582E-04^{\circ}$ 

# 1d-2) Dispersion interaction: benzene dimer

# SCC-DFTB w/fractional orbital occupation numbers

Fractional occupation numbers  $f_i$  of Kohn-Sham eigenstates replace integer  $n_i$ 

$$TB\text{-eigenvalue equation} \qquad \sum_{\nu} c_{\nu i} \left( H_{\mu\nu} - \varepsilon_i S_{\mu\nu} \right) = 0$$

$$E_{tot} = 2 \sum_i f_i \varepsilon_i + E_{rep} + \frac{1}{2} \sum_{\alpha\beta} \gamma_{\alpha\beta} \Delta q_\alpha \Delta q_\beta$$
Finite temperature approach (Mermin free energy  $E_{Mermin}$ )
M. Weinert, J. W. Davenport, *Phys. Rev. B* 45, 13709 (1992)
$$f_i = \frac{1}{\exp\left[ (\varepsilon_i - \mu) / k_B T_e \right] + 1} \qquad T_e: \text{ electronic temperature} \\ S_e: \text{ electronic entropy} \\ 0 \le f_i \le 1$$

$$E_{Mermin} = E_{tot} - T_e S_e \qquad S_e = -2k_B \sum_{i}^{\infty} f_i \ln f_i + (1 - f_i) \ln (1 - f_i)$$
Atomic force
$$\frac{r}{F_{\alpha}} = -2 \sum_{i} f_i \sum_{\mu\nu} c_{\mu i} c_{\nu i} \left[ \frac{\partial H_{\alpha}^0}{\partial R_{\alpha}} - \left( \varepsilon_i - \frac{H_{\mu\nu}^1}{S_{\mu\nu}} \right) \frac{\partial S_{\mu\nu}}{\partial R_{\alpha}} \right] - \Delta q_{\alpha} \sum_{\varepsilon}^N \frac{\partial \gamma_{\alpha\varepsilon}}{\partial R_{\alpha}} \Delta q_{\varepsilon} - \frac{\partial E_{rep}}{\partial R_{\alpha}}$$
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## 2b) Oxidation of (5,5) SWCNT

Directories: nanotube/inputs/pristine & oxidized & o\_atom

Important: Need to use electronic temperature  $T_e$  = 1000 K or similar to obtain SCC convergence

O Atom: SCC-DFTB energy: -3.0982859370 ha SWCNT: optimized SCC-DFTB energy: -128.1960247645 ha Oxidized SWCNT: optimized SCC-DFTB energy: -131.4584



Binding energy BE = -[E(SWCNT+O) – E(SWCNT) – E(O)] = 103.0 kcal/mol

#### 2c) Fe38 geometry optimization



Initial geometry:Total Energy:-55.2723102833 HTotal Mermin free energy:-55.7289532045 HMaximal force component:0.736021E-01

**Optimized geometry, T<sub>e</sub>=1500K, 144 geom.** updates:

Bonus: Calculate binding energy of C2 and Sc atoms:

BE = -[E(Sc2C2@C82) - E(C82) - 2E(Sc) - E(C2)]

 Total Energy:
 -55.6471927268 H

 Total Mermin free energy:
 -56.0689013490 H

 Maximal force component:
 0.575369E-04

# 2d) MO plots of C<sub>82</sub> and Sc<sub>2</sub>C<sub>2</sub>@C<sub>82</sub>

Directories: **C82** and **Sc2C2C82**, subdirectories input and input/orbital in both Important: Need to use electronic temperature  $T_e = 1000$  K or similar to obtain SCC convergence

Write MO's in dftb\_in.hsd: Options = { WriteDetailedXML = Yes WriteEigenvectors = Yes }

C82

# Sample "waveplot\_in.hsd"

# General options				
Options = { TotalChargeDensity = Yes TotalChargeDifference = Yes ChargeDensity = Yes RealComponent = Yes PlottedSpins = { 1 -1 } PlottedLevels = { 11 14 15 16 1 PlottedRegion = OptimalCubor	<ul> <li># Total density be plotted?</li> <li># Total density difference plotted?</li> <li># Charge density for each state?</li> <li># Plot real component of the wavefunction</li> <li>7 18 }# Levels to plot</li> <li>id {} # Region to plot</li> </ul>			
NrOfPoints = { 50 50 50 } NrOfCachedGrids = -1 Verbose = Yes }	<ul><li># Number of grid points in each direction</li><li># Nr of cached grids (speeds up things)</li><li># Wanna see a lot of messages?</li></ul>			
DetailedXML = "detailed.xml" EigenvecBin = "eigenvec.bin"	# File containing the detailed xml output of DFTB+ # File cointaining the binary eigenvecs			
<pre># Definition of the basis Basis = {     Resolution = 0.01     # Including mio-0-1.hsd. (If you use a set, which depends on other sets,     # the wfc.*.hsd files for each required set must be included in a similar     # way.)     &lt;&lt;+ "wfc.mio-0-1.hsd" }</pre>				

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# Visualize the MOs with VMD

S	elected Molecu	le	
0: wp-1-1-11-real.cube			
Create Rep		Delete Rep	
Style	Color	Selection	
CPK	Name	all	
Isosurface	ColorID 0	<volume></volume>	
Isosurface	ColorID 1	<volume></volume>	
	Selected Atoms	;	
all			
Drow style Role		on / Doriodic	
Coloring Meth	od	Material	
1			
Drawing Meth	od		
Isosurface	<b>-</b>	Default	
Range -0.196	0.1960 Vol <mark>vol</mark> i	D: C:/Docurr	
Isovalue	0.01		
Step 📢 🚺 1	)) Draw Sol	id Surface 💌	
Size 🕊 🌓 1	)) Show Iso:	surface 💌	
Apply Changes Apply			

#### Note: Run these commands on your Terminal!

•Start VMD: Run All Programs/University-of-Illinois/ VMD/VMD 1.8.7

•File  $\rightarrow$ New Molecule, then load the the desired cube file from (follow the procedure below)

-Browse

-My computer

- -Z: drive (long name with Z: at the end, double-click)
- -terminal\_<color>\_<M> (see label on your terminal)
- -C82 (or Er2C2C82) (double-click)
- -input (double-click)
- -orbital (double-click)

-select desired cube file

Note: the location of your files will depend on the directory in which you ran the DFTB+ and waveplot programs

# Visualize the MOs with VMD

Graphical Representations			
s	elected Molecu	lle	
0: wp-1-1-11-re	al.cube	▼	
Create Dep	1	Delete Dep	
	J		
Style	Color	Selection	
CPK Usosurface	Name ColorID 0	all	
Isosurface	ColorID 1	<volume></volume>	
I	Selected Atom	S	
all			
Draw style   Sele	ections   Trajec	tory Periodic	
Coloring Meth	od	Material	
ColorID		oaque 🔻	
Drawing Moth	,,		
Isosurface		Default	
Range -0.196	).1960 Vol vo	IO: C:/Docum	
Isovalue	0.01		
Step 🕊 🌓 1	)) Draw So	lid Surface 💌	
Size 📢 🕴 1	) Show Isc	osurface 💌	
	Apply C Automa	hanges tically Apply	

- •Graphics  $\rightarrow$  Representations, then choose the drawing method (here CPK)
- •Push the "Create Rep" button

•Change the drawing method "Isosurface", then change the draw type from Points to Solid surfaces and show type to Isosurface

•Change the coloring method from Name to Color ID (color 0:blue)

•Set the Isovalue (positive number; for example, 0.01 etc.)

•Perform the same thing for negative Isovalue (for example, -0.01, color 1: red)

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# 2d) MO plots of C<sub>82</sub> and Sc<sub>2</sub>C<sub>2</sub>@C<sub>82</sub>



#### 3a) H2O NVE SCC-DFTB/MD simulation



#### 3a) H2O NVE SCC-DFTB/MD simulation



Interpret your results. How are T and V related?

exit

#### 3b) H2O NVT SCC-DFTB/MD simulation



#### 3b) H2O NVT SCC-DFTB/MD simulation

dftb\_in.hsd:



E is NOT conserved in NVT simulations, but T is: 
$$\frac{1}{2}$$
 mv<sup>2</sup> =  $3/2k_{b}T$ 

$$\delta = \left[1 + \frac{\Delta t}{\tau_T} \left\{ \frac{T_0}{T(t - \Delta t)} - 1 \right\} \right]^{1/2} , \quad v_i = \delta v_i$$

In case of one water molecule, 1 au = 105258.2 K 300

awk '{print \$1\*105258.2}' t.dat > temp.dat

Interpret your results. Try different values of the coupling constant  $\tau_{\tau}$ . What is the influence of  $\tau_{\tau}$ ?

