Part I

Density-Functional Tight-Binding (DFTB) as fast approximate DFT method







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Workshop at IACS April 8 & 11, 2016

Density-Functional Tight-Binding Part I

- 1. Tight-Binding
- 2. Density-Functional Tight-Binding (DFTB)

Part II

- 3. Bond Breaking in DFTB
- 4. Extensions
- 5. Performance and Applications

Density-Functional Tight-Binding

Part I

1. Tight-Binding

2. Density-Functional Tight-Binding (DFTB)

Part II

- 3. Bond Breaking in DFTB
- 4. Extensions
- 5. Performance and Applications

1. Tight-Binding

Resources

- 1. http://www.dftb.org
- **2. DFTB** Porezag, D., T. Frauenheim, T. Köhler, G. Seifert, and R. Kaschner, *Construction of tight-binding-like potentials on the basis of density-functional theory: application to carbon.* Phys. Rev. B, 1995. **51**: p. 12947-12957.
- **3. DFTB** Seifert, G., D. Porezag, and T. Frauenheim, *Calculations of molecules, clusters, and solids with a simplified LCAO-DFT-LDA scheme.* Int. J. Quantum Chem., 1996. **58**: p. 185-192.
- **4. SCC-DFTB** Elstner, M., D. Porezag, G. Jungnickel, J. Elsner, M. Haugk, T. Frauenheim, S. Suhai, and G. Seifert, *Self-consistent-charge density-functional tight-binding method for simulations of complex materials properties*. Phys. Rev. B, 1998. **58**: p. 7260-7268.
- **5. SCC-DFTB-D** Elstner, M., P. Hobza, T. Frauenheim, S. Suhai, and E. Kaxiras, *Hydrogen bonding and stacking interactions of nucleic acid base pairs: A density-functional-theory based treatment.* J. Chem. Phys., 2001. **114**: p. 5149-5155.
- **6. SDFTB** Kohler, C., G. Seifert, U. Gerstmann, M. Elstner, H. Overhof, and T. Frauenheim, *Approximate density-functional calculations of spin densities in large molecular systems and complex solids.* Phys. Chem. Chem. Phys., 2001. **3**: p. 5109-5114.
- **7. DFTB3** Gaus, M.; Cui, C.; Elstner, M. *DFTB3: Extension of the Self-Consistent-Charge Density-Functional Tight-Binding Method (SCC-DFTB).* J. Chem. Theory Comput., 2011. **7**: p. 931-948.

Implementations

DFTB+
DF
TB

Standalone fast and efficient DFTB implementation with several useful extensions of the original DFTB method. It is developed at the Bremen Center for Computational Materials Science (Prof. Frauenheim, Balint Aradi). Based on previous DYLAX code. Free for non-commercial use.

DFTB+/Accelrys

DFTB+ as part of Accelrys' Materials Studio package, providing a user friendly graphical interface and the possibility to combine DFTB with

other higher or lower level methods.

deMon DFTB integrated in the ab initio DFT code deMon (Thomas Heine)

GAUSSIAN G09 DFTB in the Gaussian code (Keiji Morokuma)

AMBER Amber is a package of molecular simulation programs distributed by

UCSF, developed mainly for biomolecular simulations. The current

version of Amber includes QM/MM. (Marcus Elstner et al.)

CHARMm (Chemistry at HARvard Macromolecular Mechanics) (Qiang

Cui.)

ADF DFTB integrated in the Amsterdam Density Functional (ADF) program

suite. (Thomas Heine)

GAMESS-US DFTB1/2/3 and FMO2-DFTB1/2/3 (Yoshio Nishimoto, Dmitri Fedorov,

Stephan Irle)

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1. Tight-Binding

Tight-Binding

- Tight binding (TB) approaches work on the principle of treating electronic wavefunction of a system as a superposition of atom-like wavefunction (known to chemists as LCAO approach)
- Valence electrons are tightly bound to the cores (not allowed to delocalize beyond the confines of a minimal LCAO basis)
- Semi-empirical tight-binding (SETB): Hamiltonian Matrix elements are approximated by analytical functions (no need to compute integrals)
- TB energy for N electrons, M atoms system:

$$E_{\text{TB}} = \sum_{i}^{N} \epsilon_{i} + \frac{1}{2} \sum_{j \neq k}^{M} v^{j,k} (|R_{j} - R_{k}|)$$

 This separation of one-electron energies and interatomic distancedependent potential v^{j,k} constitutes the TB method

Tight-Binding

• ε_{ι} are eigenvalues of a Schrodinger-like equation

$$\epsilon_i \Psi_i(\vec{r}) = \left[-\frac{1}{2} \Delta + V(\vec{r}) \right] \Psi_i(\vec{r})$$

• solved variationally using atom-like (minimum, single-zeta) AO basis set, leading to a secular equation:

$$|\underline{H} - \epsilon \underline{S}| = 0$$

where \underline{H} and \underline{S} are Hamiltonian and overlap matrices in the basis of the AO functions. In orthogonal TB, $\underline{S} = \underline{1}$ (overlap between atoms is neglected)

 <u>H</u> and <u>S</u> are constructed using nearest-neighbor relationships; typically only nearest-neighbor interactions are considered: Similarity to extended Hückel method

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1. Tight-Binding

Extended Huckel (EHT) Method

 Based on approximation by M. Wolfsberg and L. J. Helmholz (1952)

$$HC_i = \epsilon_i SC_i$$

- H Hamiltonian matrix constructed using nearest neighbor relationships
- C_i column vector of the i-th molecular orbital coefficients
- ε_i orbital energy
- S overlap matrix
- $H_{\mu\mu}$ choose as a constant valence shell ionization potentials
- $H_{\mu\nu} = K S_{\mu\nu} (H_{\mu\mu} + H_{\nu\nu})/2$
- K Wolfsberg Helmholz constant, typically 1.75

Categories of TB approaches

TB models can conditionally be divided:

- Depending on how we treat the overlap matrix S:
 - Orthogonal TB $\mathbf{H}\Psi = E\Psi$

Non-orthogonal TB

$$H\Psi = ES\Psi$$

- Depending on how we derive the parameters:
 - Ab initio TB (derive from DFT)
- Empirical (fit to experiments, or ab initio results)
- Depending on how we treat the charge self-consistency problem:
 - Non-self-consistent
- Self-consistent

Important: in all modifications we can use the Hellmann-Feynman theorem

$$\mathbf{f}_I = -\langle \psi \mid \frac{\partial H}{\partial \mathbf{R}_I} \mid \psi \rangle$$

Source: http://beam.acclab.helsinki.fi/~akrashen/esctmp.html

1. Tight-Binding

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Slater-Koster (SK) Approximation (I)

The key moment is how we calculate one-electron integrals

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$$h_{pq} = \langle p \mid h \mid q \rangle = \int d^3r \chi_p^*(\mathbf{r}) \left[-\frac{1}{2} \nabla^2 - \sum_{\mathbf{p}} \frac{\mathbf{Z_n}}{|\mathbf{R_n} - \mathbf{r}|} \right] \chi_{\mathbf{q}}(\mathbf{r})$$

Note that the matrix element depends

on the positions of ALL atoms

Instead of introducing basis functions, e.g., Slater-like for the radial part and s,p,d for the angular part, then doing the integrals, we introduce an analytical function which depends on R_{AB} only times another function which depends on the overlap of the angular functions localized at the atoms

<< Return to Cited Reference Index Title: SIMPLIFIED LCAO METHOD FOR THE PERIODIC POTENTIAL PROBLEM Author(s): SLATER JC, KOSTER GF Source: PHYSICAL REVIEW 94 (6): 1498-1524 1954 Document Type: Article Language: English Cited References: 32 Times Cited: 2415 FIND RELATED RECORDS () Publisher: AMERICAN PHYSICAL SOC, ONE PHYSICS ELLIPSE, COLLEGE PK, MD Subject Category: PHYSICS, MULTIDISCIPLINARY

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 R_{AB}

Source: http://beam.acclab.helsinki.fi/~akrashen/esctmp.html

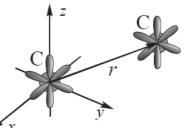
SK Approximation (II)



Let's consider two atoms with s and p atomic orbitals (e.g. Carbon)

Atom 2:
$$|s_2\rangle$$
, $|x_2\rangle$, $|y_2\rangle$, $|z_2\rangle$

$$h_{pq} = \langle p \mid h \mid q \rangle = \int d^3 r \chi_p^*(\mathbf{r}) \left[-\frac{1}{2} \nabla^2 - \sum_{\mathbf{n}} \frac{\mathbf{Z_n}}{|\mathbf{R_n} - \mathbf{r}|} \right] \chi_{\mathbf{q}}(\mathbf{r})$$

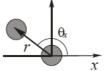


• Diagonal elements: Assume we know s and p energies $(E_s E_p)$ for isolated single atom; we further assume that these "on-site" energies are the same when the bonds are formed $h_{qq} = E_s = \langle q \mid H \mid q \rangle$ q is the sorbital

$$h_{qq} = E_p = \langle q \mid H \mid q \rangle$$
 q is any of the p orbitals

Off-diagonal elements:

s-s: Assume that for s-s atomic orbitals, $h_{p_sq_s}=s(r)E_{s,s}=s(r)V_{ss}$ s(r) is an analytical function which depends on r only



 V_{ss} is a parameter chosen to fit the results for a reference system

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Source: http://beam.acclab.helsinki.fi/~akrashen/esctmp.html

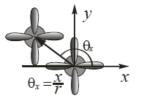
1. Tight-Binding

SK Approximation (III)

Off-diagonal elements:

s-p: Assume that for s-p atomic orbitals,

$$h_{q_sq_{p_x}} = s(r)E_{s,x} = s(r)V_{sp\sigma}l$$

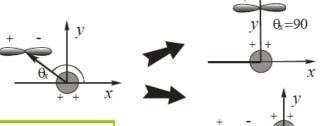


s(r) is the same analytical function

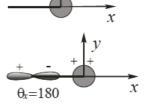
 $l = \cos(\theta_x) = x/r$ is the directional cosine of the vector between the atoms

 $V_{sp\sigma}$ is another parameter chosen to fit the results for a reference system

l reflects the anisotropy of the orbitals; note that p functions have a positive and negative part



Two-center one electron integrals are sometimes called hopping integrals



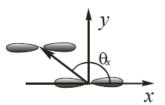
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Source: http://beam.acclab.helsinki.fi/~akrashen/esctmp.html

SK Approximation (IV)

Off-diagonal elements:

p-p: For
$$p_x$$
- p_x atomic orbitals, the matrix element is
$$h_{p_xp_x}=s(r)E_{x,x}=s(r)[l^2V_{pp\sigma}+(1-l^2)V_{pp\pi}]$$



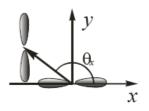
Analogously, p_x - p_y matrix element is

$$h_{p_x p_y} = s(r)E_{x,y} = s(r)[lmV_{pp\sigma} - lmV_{pp\pi}]$$

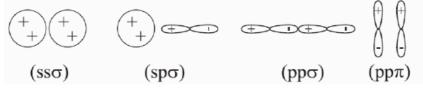
 p_z - p_z matrix element is

$$h_{p_z p_z} = s(r) E_{z,z} = s(r) [n^2 V_{pp\sigma} + (1 - n^2) V_{pp\pi}]$$

 $l = x/r, \ m = y/r, \ n = z/r$



• Four non-equivalent fundamental ontegrals between s and p atomic orbitals



p-d: We can derive similar formulas for d orbitals as well

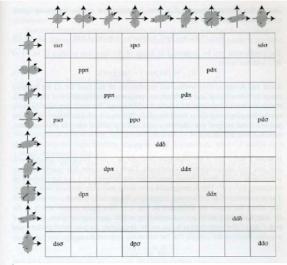
Source: http://beam.acclab.helsinki.fi/~akrashen/esctmp.html

1. Tight-Binding

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SK Tables

So, all together, the integral look as follow:



Independent two center hopping integrals between atoms on the z-axis (from M. Finnis, "Interatomic forces in Condensed Matter)

 $E_{x,x} = I^2 V_{pp\sigma} + (1 - I^2) V_{pp\sigma}$ $E_{x,y} = ImV_{pp\sigma} - ImV_{pp\pi}$ $E_{x,z} = InV_{pp\sigma} - InV_{pp\sigma}$ $E_{s,xy} = 3^{1/2} Im V_{sd\sigma}$ $E_{a,v^2-v^2} = \frac{1}{2}3^{1/2}(I^2 - m^2)V_{add}$ $E_{s,3e^2-r^2} = [n^2 - \frac{1}{2}(l^2 + m^2)]V_{sdo}$ $E_{x,xy} = 3^{1/2} I^2 m V_{pd\sigma} + m(1 - 2I^2) V_{pd\pi}$ $E_{x,yz} = 3^{1/2} Imn V_{pd\sigma} - 2 Imn V_{pd\sigma}$ $E_{x,zx} = 3^{1/2} l^2 n V_{\text{pd}s} + n(1 - 2l^2) V_{\text{pd}s}$ $E_{x,x^2-y^2} = \frac{1}{2}3^{1/2}/(I^2 - m^2)V_{pdr} + I(1 - I^2 + m^2)V_{pdr}$ $E_{\nu,x^2-y^2} = \frac{1}{2}3^{1/2}m(I^2 - m^2)V_{\text{pde}} - m(1 + I^2 - m^2)V_{\text{pdx}}$ $E_{z,x^2-y^2} = \frac{1}{2}3^{1/2}n(l^2-m^2)V_{pd\sigma} - n(l^2-m^2)V_{pd\pi}$ $E_{x,3z^2-r^2} = I[n^2 - \frac{1}{2}(I^2 + m^2)]V_{pd\sigma} - 3^{1/2}In^2V_{pdz}$ $E_{\gamma,3z^2-r^2} = m[n^2 - \frac{1}{2}(l^2 + m^2)]V_{pde} - 3^{1/2}mn^2V_{pde}$ $E_{z,3z^2-r^2} = n[n^2 - \frac{1}{2}(f^2 + m^2)]V_{\rm pd\sigma} + 3^{1/2}n(f^2 + m^2)V_{\rm pd\sigma}$ $E_{xy.xy} = 3l^2m^2V_{\rm dd\sigma} + (l^2 + m^2 - 4l^2m^2)V_{\rm dd\pi} + (n^2 + l^2m^2)V_{\rm dd\theta}$ $E_{xy,yx} = 3lm^2nV_{dd\sigma} + ln(1 - 4m^2)V_{dd\pi} + ln(m^2 - 1)V_{dd\pi}$ $E_{xy,zx} = 3l^2 mn V_{dd\sigma} + mn(1 - 4l^2) V_{dd\pi} + mn(l^2 - 1) V_{dd\pi}$ $E_{xy,x^2-y^2} = \frac{3}{2}Im(I^2 - m^2)V_{dd\sigma} + 2Im(m^2 - I^2)V_{dd\pi} + \frac{1}{2}Im(I^2 - m^2)V_{dd\delta}$ $\begin{aligned} E_{yz,x^2-y^2} &= \frac{3}{2}mn(l^2 - m^2)V_{\text{dd}\sigma} - mn[1 + 2(l^2 - m^2)]V_{\text{dd}\pi} \\ &+ mn[1 + \frac{1}{2}(l^2 - m^2)]V_{\text{dd}\sigma} \end{aligned}$
$$\begin{split} E_{2x,x^2-y^2} &= \tfrac{3}{2} n! (l^2 - m^2) V_{\text{dd}\sigma} + n! [1 - 2(l^2 - m^2)] V_{\text{dd}\pi} \\ &- n! [1 - \tfrac{1}{2} (l^2 - m^2)] V_{\text{dd}\theta} \end{split}$$
 $E_{xy,3z^2-r^2} = 3^{1/2} lm [n^2 - \frac{1}{2}(l^2 + m^2)] V_{dd\sigma} - 3^{1/2} 2lmn^2 V_{dd\tau} + \frac{1}{2} 3^{1/2} lm (1 + n^2) V_{dd\delta}$ $E_{yz,3z^2-r^2} = 3^{1/2}mn(r^2 - \frac{1}{2}(l^2 + m^2))V_{\text{odd}r} + 3^{1/2}mn(l^2 + m^2 - r^2)V_{\text{odd}r}$

Source: http://beam.acclab.helsinki.fi/~akrashen/esctmp.html

Density-Functional Tight-Binding

Part I

1. Tight-Binding

2. Density-Functional Tight-Binding (DFTB)

Part II

- 3. Bond Breaking in DFTB
- 4. Extensions
- 5. Performance and Applications

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DFTB

Taken from Oliviera, Seifert, Heine, Duarte, *J. Braz. Chem. Soc.* 20, 1193-1205 (2009)



Thomas Heine



Helio Duarte

J. Braz. Chem. Soc., Vol. 20, No. 7, 1193-1205, 2009. Printed in Brazil - ©2009 Sociedade Brasileira de Química 0103 - 5053 \$6.00+0.00

Density-Functional Based Tight-Binding: an Approximate DFT Method

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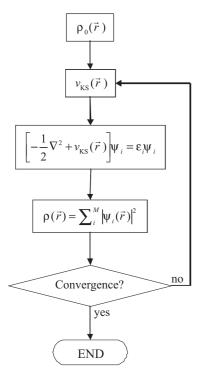
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Density Functional Theory (DFT)



1998



Walter Kohn/John A. Pople at convergence:

$$\begin{split} E\left[\rho\right] &= \sum_{i=1}^{M} n_{i} \left\langle \psi_{i} \right| - \frac{1}{2} \nabla^{2} + v_{ext} \binom{\mathbf{r}}{r} + \int \frac{\rho \binom{\mathbf{r}}{r'}}{\left| \overset{\mathbf{r}}{r} - \overset{\mathbf{r}}{r'} \right|} d^{3} r' \left| \psi_{i} \right\rangle \\ &+ E_{xc} \left[\rho\right] - \frac{1}{2} \iint \frac{\rho \binom{\mathbf{r}}{r} \rho \binom{\mathbf{r}}{r'}}{\left| \overset{\mathbf{r}}{r} - \overset{\mathbf{r}}{r'} \right|} d^{3} r d^{3} r' + \frac{1}{2} \sum_{\substack{\alpha, \beta = 1 \\ \alpha \neq \beta}}^{N} \frac{Z_{\alpha} Z_{\beta}}{\left| \overset{\mathbf{r}}{R_{\alpha}} - \overset{\mathbf{r}}{R_{\beta}} \right|} \\ &= \sum_{i=1}^{M} n_{i} \varepsilon_{i} + E_{rep} \end{split}$$

Various criteria for convergence possible:

- Electron density
- Potential
- Orbitals
- Energy
- · Combinations of above quantities

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DFTB

Foulkes + Haydock Ansatz

Phys. Rev. B, 39, 12520 (1989)

$$\begin{split} \rho(\vec{r}) &= \rho_{0}(\vec{r}) + \delta\rho(\vec{r}) \\ E[\rho_{0} + \delta\rho] &= \sum_{i}^{M} n_{i} \left\langle \Psi_{i} \middle| -\frac{1}{2} \nabla^{2} + v_{\text{ext}}(\vec{r}) + \int \frac{\rho_{0}'}{\left| \vec{r} - \vec{r}' \right|} d\vec{r}' + v_{\text{xc}}[\rho_{0}] \middle| \Psi_{i} \right\rangle \\ &- \frac{1}{2} \iint \frac{\rho_{0}' \left(\rho_{0} + \delta\rho \right)}{\left| \vec{r} - \vec{r}' \right|} d\vec{r} d\vec{r}' - \int v_{\text{xc}}[\rho_{0}] (\rho_{0} + \delta\rho) d\vec{r} \\ &+ \frac{1}{2} \iint \frac{\delta\rho' \left(\rho_{0} + \delta\rho \right)}{\left| \vec{r} - \vec{r}' \right|} d\vec{r} d\vec{r}' + E_{\text{xc}}[\rho_{0} + \delta\rho] + E_{\text{nn}} \end{split}$$

Self-consistent-charge density-functional tight-binding (SCC-DFTB)

M. Elstner et al., Phys. Rev. B 58 7260 (1998)

Approximate density functional theory (DFT) method!

Second order-expansion of DFT energy in terms of reference density ρ_0 and charge fluctuation ρ_1 ($\rho = \rho_0 + \rho_1$) yields:

$$E[\rho] = \underbrace{\sum_{i}^{\text{valence orbitals}}}_{1} n_{i} \langle \phi_{i} | \hat{H}[\rho_{0}] | \phi_{i} \rangle + \underbrace{\sum_{i}^{\text{core orbitals}}}_{2} n_{i} \langle \phi_{i} | \hat{H}[\rho_{0}] | \phi_{i} \rangle + \underbrace{E_{\text{xc}}[\rho_{0}]}_{3} - \underbrace{\frac{1}{2} \int_{\mathbf{R}^{3}} \rho_{0} V_{H}[\rho_{0}]}_{4} - \underbrace{\int_{\mathbf{R}^{3}} \rho_{0} V_{\text{xc}}[\rho_{0}]}_{5} + \underbrace{\frac{1}{2} \int_{\mathbf{R}^{3}} \rho_{1} V_{H}[\rho_{1}]}_{7} + \underbrace{\frac{1}{2} \int_{\mathbf{R}^{3}} \frac{\delta^{2} E_{\text{xc}}}{\delta \rho_{1}^{2}} \Big|_{\rho_{0}} \rho_{1}^{2} + o(3)$$

Density-functional tight-binding (DFTB) method is derived from terms 1-6

Self-consistent-charge density-functional tight-binding (SCC-DFTB) method is derived from terms 1-8

DFTB

DFTB and SCC-DFTB methods

$$E^{\text{DFTB}} = \underbrace{\sum_{i}^{\text{valence orbitals}}}_{\text{term 1}} n_{i} \varepsilon_{i} + \underbrace{\frac{1}{2} \sum_{A \neq B}^{\text{atoms}} E_{\text{rep}}^{AB}}_{\text{terms 2-6}}$$

$$E^{\text{SCC-DFTB}} = \underbrace{\sum_{i}^{\text{valence orbitals}}}_{\text{term 1}} n_{i} \varepsilon_{i} + \underbrace{\frac{1}{2} \sum_{A,B}^{\text{atoms}} \gamma_{AB} \Delta q_{A} \Delta q_{B}}_{\text{terms 7-8}} + \underbrace{\frac{1}{2} \sum_{A \neq B}^{\text{atoms}} E_{\text{rep}}^{AB}}_{\text{terms 2-6}}$$

where

- \triangleright n_i and ε_i occupation and orbital energy of the i^{th} Kohn-Sham eigenstate
- \triangleright E_{rep} distance-dependent diatomic repulsive potentials
- $ightharpoonup \Delta q_A$ induced charge on atom A
- γ_{AB} distance-dependent charge-charge interaction functional; obtained from chemical hardness (IP EA)

DFTB method

- Repulsive diatomic potentials replace usual nuclear repulsion energy
- lacktriangledown Reference density ho_0 is constructed from atomic densities

$$\rho_0 = \sum_{A} \rho_0^A$$

 $\rho_0 = \sum_A^{\text{aroms}} \rho_0^A$ * Kohn-Sham eigenstates ϕ_i are expanded in Slater basis of valence pseudoatomic orbitals χ_i

$$\phi_i = \sum_{i=1}^{AO} c_{\mu i} \chi_{\mu}$$

The DFTB energy is obtain by solving a generalized DFTB eigenvalue problem with Ho computed by atomic and diatomic DFT

$$\mathbf{H}^{0}\mathbf{C} = \mathbf{SC}\varepsilon \qquad \text{with} \quad S_{\mu\nu} = \left\langle \chi_{\mu} \middle| \chi_{\nu} \right\rangle$$

$$H_{\mu\nu}^{0} = \left\langle \chi_{\mu} \middle| \hat{H} \middle[\rho_{0}^{M}, \rho_{0}^{N} \middle] \middle| \chi_{\nu} \right\rangle$$

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DFTB

Approximations in the DFTB Hamiltonian

Traditional DFTB concept: Hamiltonian matrix elements are approximated to two-center terms. The same types of approximations are done to E_{ren}.

$$H_{\mu\nu}^{0} = \begin{cases} \varepsilon_{\mu}^{\text{neutral free atom}} & \text{if } \mu = \nu \\ \left\langle \varphi_{\mu}^{\alpha} \middle| \hat{T} + V_{0}^{\alpha} + V_{0}^{\beta} \middle| \varphi_{\nu}^{\beta} \right\rangle & \text{if } \alpha \neq \beta \\ 0 & \text{otherwise.} \end{cases}$$
 From Elstner et al., PRB 1998

$$V_{eff} [\rho_0] \approx V_{eff} [\rho_A + \rho_B]$$
 (Density superposition)
$$V_{eff} [\rho_0] \approx V_{eff} [\rho_A] + V_{eff} [\rho_B]$$
 (Potential superposition)

Both approximations are justified by the screening argument: Far away, neutral atoms have no Coulomb contribution.



SCC-DFTB matrix elements

LCAO ansatz of wave function

$$\Psi_{i} = \sum_{\nu} c_{\nu}^{i} \underline{\phi_{\nu} (\mathbf{r} - \mathbf{R}_{\alpha})}$$
pseudoatomic orbita

variational

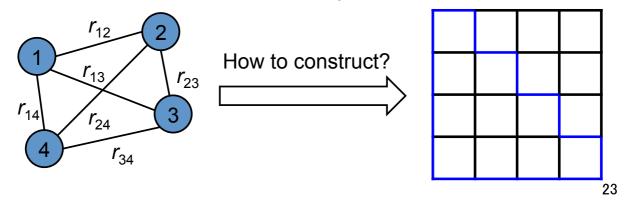
secular equations $\sum_{\nu} c_{\nu}^{i} (H_{\mu\nu} - \varepsilon_{i} S_{\mu\nu}) = 0$

Hamiltonian Overlap

pre-computed parameter

- •Reference Hamiltonian H⁰
- •Overlap integral S_{uv}
- √two-center approximation
- ✓ nearest neighbor off-diagonal elements only (choice of cutoff values)

Atom 1 – 4 are the same atom & have only s shell



SCC-DFTB matrix elements

LCAO ansatz of wave function

$$\Psi_{i} = \sum_{v} c_{v}^{i} \underline{\phi_{v}(\mathbf{r} - \mathbf{R}_{\alpha})}$$
pseudoatomic orbita

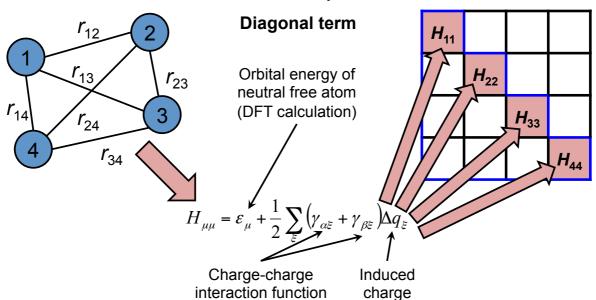
variational principle

secular equations $\sum_{a^i} (u - c \cdot S^i)$

 $\sum_{v} c_{v}^{i} \left(H_{\mu v} - \varepsilon_{i} S_{\mu v} \right) = 0$

Atom 1 - 4 are the same atom & have only s shell

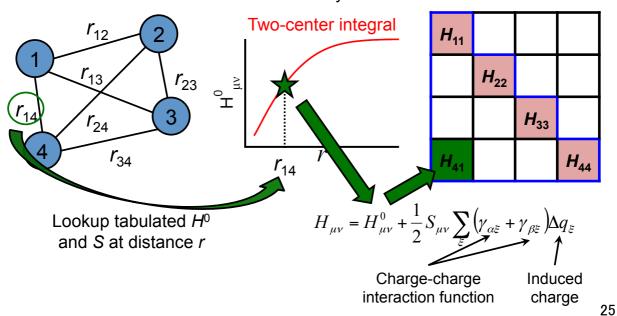
Hamiltonian Overlap



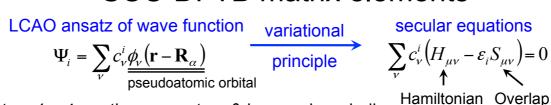
SCC-DFTB matrix elements

LCAO ansatz of wave function
$$\Psi_i = \sum_{\nu} c_{\nu}^i \underline{\phi_{\nu}(\mathbf{r} - \mathbf{R}_{\alpha})}_{\text{pseudoatomic orbital}} \text{ principle}$$
 secular equations
$$\sum_{\nu} c_{\nu}^i (H_{\mu\nu} - \varepsilon_i S_{\mu\nu}) = 0$$
 Hamiltonian Overlap

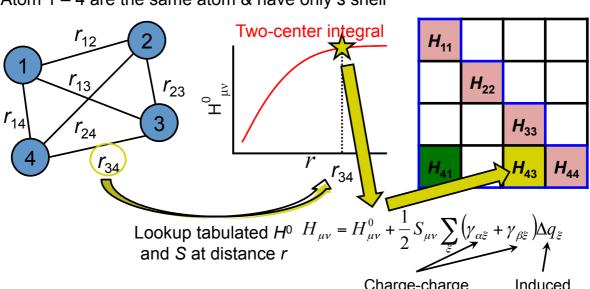
Atom 1 – 4 are the same atom & have only s shell



SCC-DFTB matrix elements



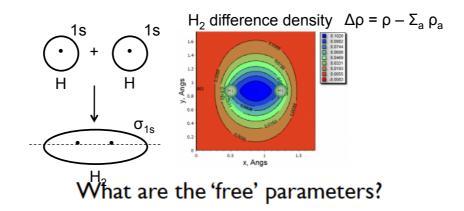
Atom 1 - 4 are the same atom & have only s shell



> Repeat until building off-diagonal term

Charge-charge Induced interaction function charge

DFTB parameters

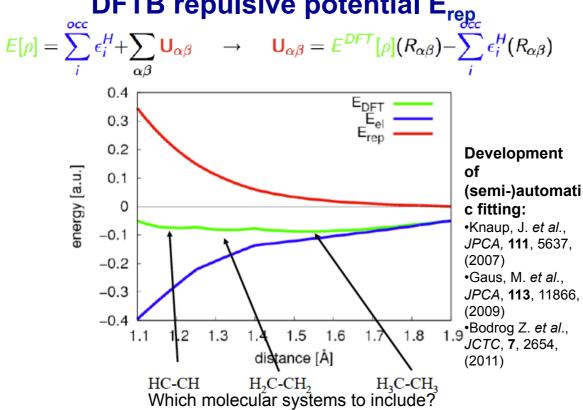


- confinement for wave-function:
- confinement for density:
- repulsive potential $U_{\alpha\beta}$:
 - 6th order polynomial (spline)
 - 20-40 DFT data points

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DFTB

DFTB repulsive potential E_{rep}



SCC-DFTB method (I)

- Additional induced-charges term allows for a proper description of charge-transfer phenomena
- Induced charge Δq_A on atom A is determined from Mulliken population analysis

$$\Delta q_A = \sum_{i}^{\text{MO}} n_i \sum_{\mu \in A} \sum_{v}^{\text{AO}} c_{\mu i} c_{v i} S_{\mu v} - q_A^0$$

Kohn-Sham eigenenergies are obtained from a generalized, self-consistent SCC-DFTB eigenvalue problem

$$\mathbf{HC} = \mathbf{SC}\,\varepsilon \qquad \text{with} \qquad S_{\mu\nu} = \left\langle \chi_{\mu} \middle| \chi_{\nu} \right\rangle \quad \text{and}$$

$$H_{\mu\nu} = \left\langle \chi_{\mu} \middle| \hat{H} \middle[\rho_{0}^{M}, \rho_{0}^{N} \middle] \chi_{\nu} \right\rangle + \frac{1}{2} S_{\mu\nu} \sum_{K}^{\text{atoms}} \left(\gamma_{MK} + \gamma_{NK} \right) \Delta q_{K}$$

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DFTB

SCC-DFTB method (II)

Only second-order terms (terms 7-8 on slide 16):

$$E^{2}[\rho,\rho_{0}] = \frac{1}{2} \int \int' \left(\frac{1}{|\vec{r}-\vec{r}'|} + \left. \frac{\delta^{2} E_{xc}}{\delta \rho \, \delta \rho'} \right|_{n_{0}} \right) \Delta \rho \, \Delta \rho'$$

Represent by atomic contributions:

$$\Delta \rho = \sum_{\alpha} \Delta \rho_{\alpha}$$

Monopole approximation

$$\Delta \rho_{\alpha} \approx \Delta q_{\alpha} F_{00}^{\alpha} Y_{00}$$

$$[E^2[
ho,
ho_0]pprox rac{1}{2}\sum_{lphaeta}\Delta q_lpha\Delta q_eta\int\int'\left(rac{1}{|ec r-ec r'|}+\left.rac{\delta^2 E_{
m xc}}{\delta
ho\,\delta
ho'}
ight|_{n_0}
ight)F^lpha_{00}F^eta_{00}Y^2_{00}$$

Basic assumptions:

- Only transfer of net charge between atoms
- ·Size and shape of atom (in molecule) unchanged

SCC-DFTB method (III)

$$[E^2[
ho,
ho_0]pproxrac{1}{2}\sum_{lphaeta}\Delta q_lpha\Delta q_eta\int\int'\left(rac{1}{|ec r-ec r'|}+\left.rac{\delta^2 extbf{ ilde{E}}_{xc}}{\delta
ho\,\delta
ho'}
ight|_{m{n_0}}
ight)F_{00}^lpha F_{00}^eta Y_{00}^2$$

Consider two limiting cases:

A) Large distance: $|\vec{r} - \vec{r}'| \approx R_{\alpha\beta} \rightarrow \infty$

$$E^2[
ho,
ho_0]
ightarrow rac{1}{2} \sum_{lphaeta} rac{\Delta q_lpha \Delta q_eta}{R_{lphaeta}}$$

B) 'On-site' e-e repulsion: $R_{lphaeta} o 0$, i.e. lpha=eta

$$[E^2[
ho,
ho_0]
ightarrow rac{1}{2}\sum_lpha rac{\partial^2 E^{at}_lpha}{\partial q^2_lpha} \Delta q^2_lpha = rac{1}{2}\sum_lpha U_lpha \Delta q^2_lpha$$

New (Hubbard) parameter U_{α} : computed from DFT (PBE)

DFTB

SCC-DFTB method (IV)

Derive analytical function which interpolates between these two cases

$$E^{2}[\rho,\rho_{0}] \approx \frac{1}{2} \sum_{\alpha\beta} \Delta q_{\alpha} \Delta q_{\beta} \int \int' \left(\frac{1}{|\vec{r}-\vec{r}'|} + \left. \frac{\delta^{2} E_{xc}}{\delta \rho \, \delta \rho'} \right|_{\textbf{\textit{n}}_{0}} \right) F_{00}^{\alpha} F_{00}^{\beta} Y_{00}^{2}$$

$$E^2[\rho, \rho_0] pprox rac{1}{2} \sum_{lpha eta} \Delta q_lpha \Delta q_eta \gamma_{lpha eta}$$

Klopmann-Ohno:

$$\gamma_{\alpha\beta} = \frac{1}{\sqrt{R^2 + \frac{1}{4} \left(U_{\alpha}^{-1} + U_{\beta}^{-1} \right)^2}}$$

Several possible formulations for $\gamma_{\alpha\beta}$: Mataga-Nishimoto < Klopmann-Ohno < DFTB

Elstner et al. (1998): Phys. Rev. B 58, 7260 (1998)

Gradient for the DFTB methods

The DFTB force formula

$$F_{a} = -\sum_{i}^{\text{MO}} n_{i} \sum_{\mu\nu}^{\text{AO}} c_{\mu i} c_{\nu i} \left[\frac{\partial H_{\mu\nu}^{0}}{\partial a} - \varepsilon_{i} \frac{\partial S_{\mu\nu}}{\partial a} \right] - \frac{\partial E_{\text{rep}}}{\partial a}$$

The SCC-DFTB force formula

$$F_{a} = -\sum_{i}^{\text{MO}} n_{i} \sum_{\mu\nu}^{\text{AO}} c_{\mu i} c_{\nu i} \left[\frac{\partial H_{\mu\nu}^{0}}{\partial a} - \left(\varepsilon_{i} - \frac{1}{2} \sum_{K}^{\text{atoms}} (\gamma_{MK} + \gamma_{NK}) \Delta q_{K} \right) \frac{\partial S_{\mu\nu}}{\partial a} \right] -$$

$$-\Delta q_{A} \sum_{K}^{\text{atoms}} \frac{\partial \gamma_{AK}}{\partial a} \Delta q_{K} - \frac{\partial E_{\text{rep}}}{\partial a}$$

computational effort: energy calculation 90% gradient calculation 10%

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DFTB

Spin-polarized DFTB (SDFTB)

- ❖ for systems with different ↑ and ↓ spin densities, we have
 - > total density $\rho = \rho^{\uparrow} + \rho^{\downarrow}$
 - ightharpoonup magnetization density $ho^{\rm S}=
 ho^{\uparrow}$ ho^{\downarrow}
- riangle 2nd-order expansion of DFT energy at $(
 ho_0,0)$ yields

$$E[\rho, \rho^{S}] = \underbrace{\sum_{i}^{\text{valence orbitals}}}_{1} n_{i} \langle \phi_{i} | \hat{H}[\rho_{0}] | \phi_{i} \rangle + \underbrace{\sum_{i}^{\text{core orbitals}}}_{2} n_{i} \langle \phi_{i} | \hat{H}[\rho_{0}] | \phi_{i} \rangle + \underbrace{E_{\text{xc}}[\rho_{0}]}_{3} - \underbrace{\frac{1}{2} \int_{\mathbf{R}^{3}} \rho_{0} V_{H}[\rho_{0}]}_{4} - \underbrace{\frac{1}{2} \int_{\mathbf{R}^{3}} \rho_{0} V_{H}[\rho_{0}]}_{$$

$$-\underbrace{\int_{\mathbf{R}^{3}} \rho_{0} V_{xc}[\rho_{0}]}_{5} + \underbrace{E_{nucl}}_{6} + \underbrace{\frac{1}{2} \int_{\mathbf{R}^{3}} \rho_{1} V_{H}[\rho_{1}]}_{7} + \underbrace{\frac{1}{2} \iint_{\mathbf{R}^{3}} \frac{\delta^{2} E_{xc}}{\delta \rho_{1}^{2}} \Big|_{(\rho_{0},0)} \rho_{1}^{2}}_{8} + \underbrace{\frac{1}{2} \iint_{\mathbf{R}^{3}} \frac{\delta^{2} E_{xc}}{\left(\delta \rho^{S}\right)^{2}} \Big|_{(\rho_{0},0)} (\rho^{S})^{2} + o(3)$$

The Spin-Polarized SCC-DFTB (SDFTB) method is derived from terms 1-9

Spin-polarized DFTB (SDFTB)

$$E^{\text{SDFTB}} = \underbrace{\sum_{i}^{\text{valence orbitals}} n_{i}^{\uparrow} \mathcal{E}_{i}^{\uparrow} + \sum_{i}^{\text{valence orbitals}} n_{i}^{\downarrow} \mathcal{E}_{i}^{\downarrow}}_{\text{term 1}} + \underbrace{\frac{1}{2} \sum_{A \neq B}^{\text{atoms}} \gamma_{AB} \Delta q_{A} \Delta q_{B}}_{\text{terms 7-8}} + \underbrace{\frac{1}{2} \sum_{A \neq B}^{\text{atoms}} E_{\text{rep}}^{AB}}_{\text{terms 2-6}} + \underbrace{\frac{1}{2} \sum_{A \neq B}^{\text{atoms}} \sum_{l \in A} \sum_{l \in A} \sum_{l \in A} \sum_{l \in A} p_{A l} p_{A l'} W_{A l l'}}_{\text{term 9}}$$

where p_{AI} — spin population of shell I on atom A

$$W_{A\,II'}$$
 — spin-population interaction functional $W_{AII'} = \frac{1}{2} \left(\frac{\partial \epsilon_{AI}^\dagger}{\partial n_r^\dagger} - \frac{\epsilon_{AI}^\dagger}{\partial n_r^\dagger} \right)_{\rho=0} = W_{AII}$

Spin populations p_{AI} and induced charges Δq_A are obtained from Mulliken population analysis

$$\Delta q_A = \sum_{i}^{\text{MO}} \sum_{u \in A} \sum_{v}^{\text{AO}} \left(n_i^{\uparrow} c_{ui}^{\uparrow} c_{vi}^{\uparrow} + n_i^{\downarrow} c_{ui}^{\downarrow} c_{vi}^{\downarrow} \right) S_{\mu v} - q_A^0$$

$$p_{Al} = \sum_{i}^{MO} \sum_{\mu \in A} \sum_{i}^{AO} \left(n_i^{\uparrow} c_{\mu i}^{\uparrow} c_{\nu i}^{\uparrow} - n_i^{\downarrow} c_{\mu i}^{\downarrow} c_{\nu i}^{\downarrow} \right) S_{\mu \nu}$$

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DFTB

Spin-polarized DFTB (SDFTB)

Kohn-Sham energies are obtained by solving generalized, self-consistent SDFTB eigenvalue problems

$$\mathbf{H}^{\uparrow}\mathbf{C}^{\uparrow} = \mathbf{S}\mathbf{C}^{\uparrow}\varepsilon^{\uparrow}$$

$$\mathbf{H}^{\downarrow}\mathbf{C}^{\downarrow} = \mathbf{S}\mathbf{C}^{\downarrow}\varepsilon^{\downarrow}$$

where

$$S_{\mu\nu} = \left\langle \chi_{\mu} \middle| \chi_{\nu} \right\rangle$$

$$H_{\mu\nu}^{\uparrow} = \left\langle \chi_{\mu} \middle| \hat{H} \Big[\rho_0^M, \rho_0^N \Big] \middle| \chi_{\mu} \right\rangle + \frac{1}{2} S_{\mu\nu} \sum_{K}^{\text{atoms}} \left(\gamma_{MK} + \gamma_{NK} \right) \Delta q_K + \delta_{MN} \frac{1}{2} S_{\mu\nu} \sum_{l'' \in M} \left(W_{All'} + W_{All''} \right) p_{Ml''}$$

$$H_{\mu\nu}^{\downarrow} = \left\langle \chi_{\mu} \middle| \hat{H} \middle[\rho_{0}^{M}, \rho_{0}^{N} \middle] \middle| \chi_{\mu} \right\rangle + \frac{1}{2} S_{\mu\nu} \sum_{K}^{\text{atoms}} \left(\gamma_{MK} + \gamma_{NK} \right) \Delta q_{K} - \delta_{MN} \frac{1}{2} S_{\mu\nu} \sum_{I^{"} \in M} \left(W_{All^{"}} + W_{All^{"}} \right) p_{MI^{"}}$$

M,N,K: indexing specific atoms

SCC-DFTB w/fractional orbital occupation numbers

Fractional occupation numbers f_i of Kohn-Sham eigenstates replace integer n_i

TB-eigenvalue equation $\sum c_{vi} \left(H_{\mu\nu} - \varepsilon_i S_{\mu\nu} \right) = 0$

$$E_{tot} = 2\sum_{i} f_{i} \varepsilon_{i} + E_{rep} + \frac{1}{2} \sum_{\alpha\beta} \gamma_{\alpha\beta} \Delta q_{\alpha} \Delta q_{\beta}$$

Finite temperature approach (Mermin free energy E_{Mermin})

M. Weinert, J. W. Davenport, Phys. Rev. B 45, 13709 (1992)

$$f_i = \frac{1}{\exp\left[\left(\varepsilon_i - \mu\right)/k_B T_e\right] + 1}$$

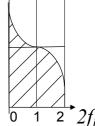
$$E_{Mermin} = E_{tot} - T_e S_e$$

$$f_{i} = \frac{1}{\exp\left[\left(\varepsilon_{i} - \mu\right)/k_{B}T_{e}\right] + 1}$$

$$T_{e}: \text{ electronic temperature } S_{e}: \text{ electronic entropy } 0 \le f_{i} \le 1$$

$$E_{Mermin} = E_{tot} - T_{e}S_{e}$$

$$S_{e} = -2k_{B}\sum_{i}^{\infty} f_{i} \ln f_{i} + (1 - f_{i}) \ln (1 - f_{i})$$



Atomic force

The force
$$F_{\alpha} = -2\sum_{i} f_{i} \sum_{\mu\nu} c_{\mu i} c_{\nu i} \left[\frac{\partial H_{\mu\nu}^{0}}{\partial R_{\alpha}} - \left(\varepsilon_{i} - \frac{H_{\mu\nu}^{1}}{S_{\mu\nu}} \right) \frac{\partial S_{\mu\nu}}{\partial R_{\alpha}} \right] - \Delta q_{\alpha} \sum_{\varepsilon}^{N} \frac{\partial \gamma_{\alpha\varepsilon}}{\partial R_{\alpha}} \Delta q_{\varepsilon} - \frac{\partial E_{\kappa ep}}{\partial R_{\alpha}}$$

DFTB

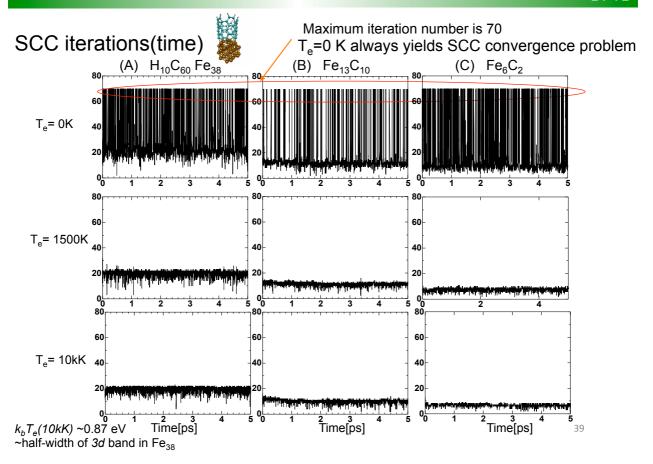
Fermi-Dirac distribution function: Energy derivative for Mermin Free Energy

M. Weinert, J. W. Davenport, Phys. Rev. B 45, 13709 (1992)

$$\begin{split} F_{\alpha}^{elect} & \equiv \vec{F}_{HF} + \vec{F}_{pulay} + \vec{F}_{charge} \\ & = \sum_{i}^{\infty} \frac{\partial}{\partial x} \left(\varepsilon_{i} f_{i} \right) = \sum_{i}^{\infty} f_{i} \frac{\partial \varepsilon_{i}}{\partial x} + \sum_{i}^{\infty} \varepsilon_{i} \frac{\partial f_{i}}{\partial x} \end{split}$$

$$\begin{split} F_{\alpha}^{\textit{elect-TS}} &\equiv \vec{F}_{HF} + \vec{F}_{\textit{pulay}} + \vec{F}_{\textit{charge}} + \vec{F}_{\textit{TS}} \\ &= \sum_{i}^{\infty} \frac{\partial}{\partial x} \left(\varepsilon_{i} f_{i} \right) + \frac{\partial \left(-T_{e} S \right)}{\partial x} \\ &= \sum_{i}^{\infty} f_{i} \frac{\partial \varepsilon_{i}}{\partial x} + \sum_{i}^{\infty} \varepsilon_{i} \frac{\partial f_{i}}{\partial x} - \sum_{i}^{\infty} \varepsilon_{i} \frac{\partial f_{i}}{\partial x} \end{split} \quad \begin{array}{c} \text{Correction term arising from Fermi distribution function cancels out} \\ \end{array}$$

$$= \sum_{i}^{\infty} f_{i} \frac{\partial \mathcal{E}_{i}}{\partial x}$$



Electronic Parameters

New Confining Potentials

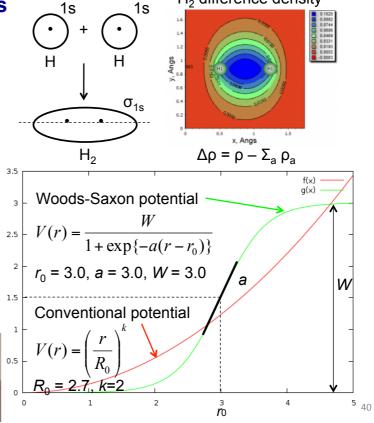
➤ Typically, electron density contracts under covalent bond formation.

➤In standard *ab initio* methods, this problem can be remedied by including more basis functions.

➤DFTB uses minimal valence basis set: the confining potential is adopted to mimic contraction

DFTB Parameterization

H₂ difference density



Henryk Witek

1). DFT band structure calculations

- •VASP 4.6
- One atom per unit cell
- •PAW (projector augmented wave) method
- •32 x 32 x 32 Monkhorst-Pack k-point sampling
- •cutoff = 400 eV
- •Fermi level is shifted to 0 eV

2). DFTB band structure fitting

- •Optimization of parameter sets for Woods-Saxon confining potential (orbital and density) and unoccupied orbital energies
- •Fixed orbital energies for electron occupied orbitals
- •Valence orbitals : [1s] for 1st row

[2s, 2p] for 2nd row

[ns, np, md] for 3rd - 6th row

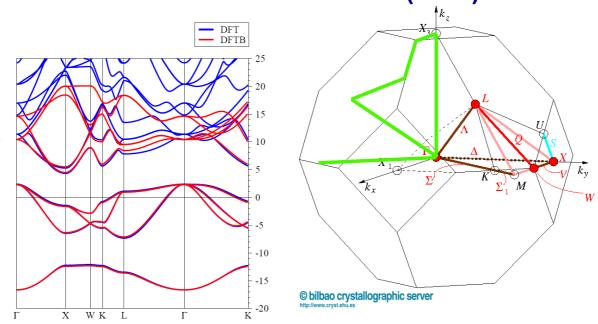
 $(n \ge 3, m = n-1 \text{ for group } 1-12, m = n \text{ for group } 13-18)$

•Fitting points : valence bands + conduction bands (depending on the system, at least including up to ~+5 eV with respect to Fermi level)

Electronic Parameters

DFTB Parameterization

Band structure for Se (FCC)

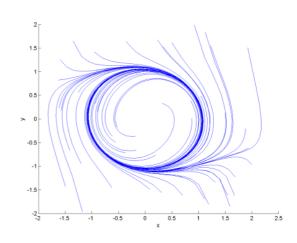


Brillouin zone

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Particle swarm optimization (PSO)

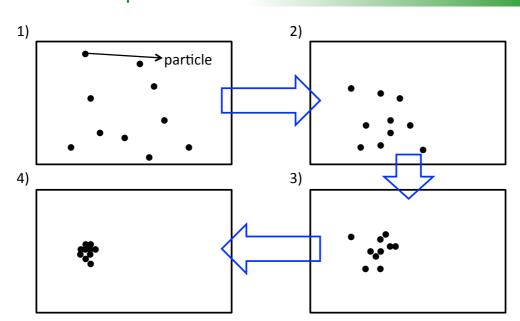




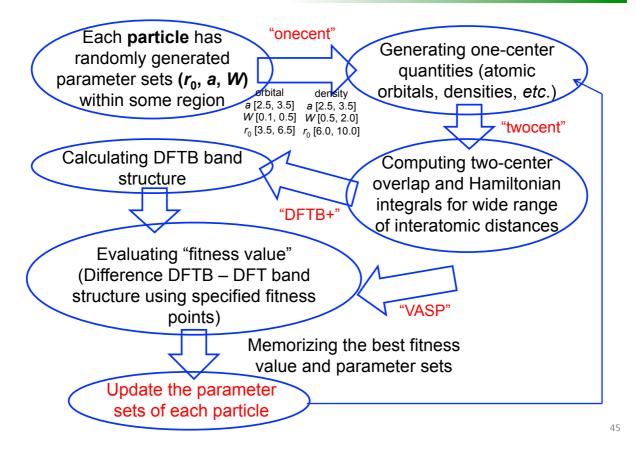
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Particle Swarm Optimization

DFTB Parameterization



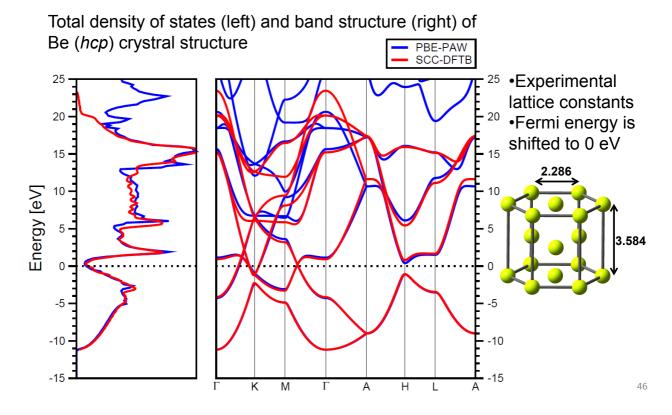
- 1) Particles (=candidate of a solution) are randomly placed initially in a target space.
- (2) 3) Position and velocity of particles are updated based on the exchange of information between particles and particles try to find the best solution.
- 4) Particles converges to the place which gives the best solution after a number of iterations.



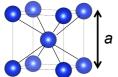
Electronic Parameters

DFTB Parameterization

Example: Be, HCP crystal structure



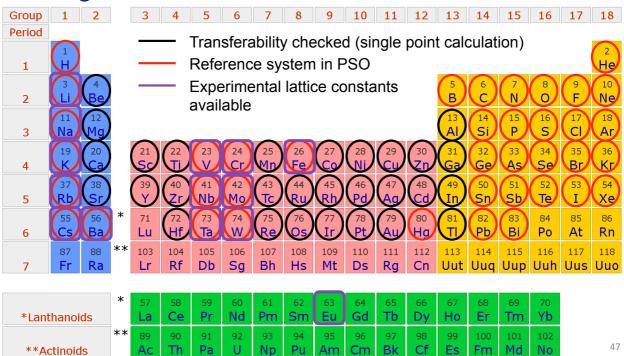
Band structure fitting for BCC crystal structures



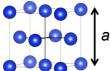
•space group No. 229

 \triangleright No POTCAR file for Z ≥ 84 in VASP

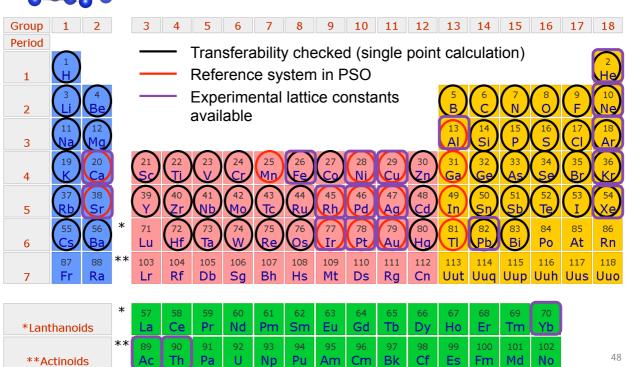
•1 lattice constant (a)



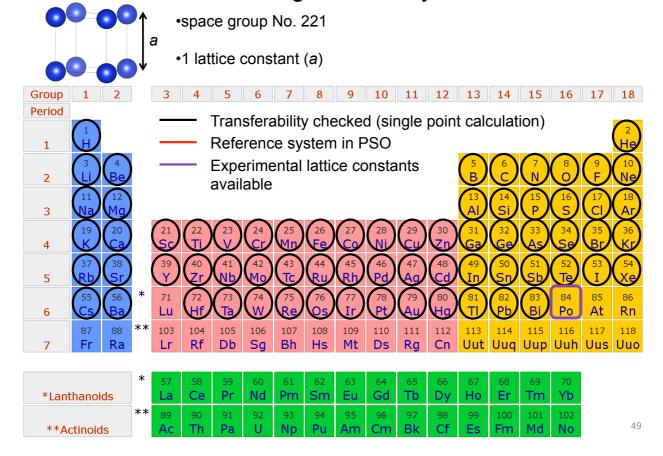
Band structure fitting for FCC crystal structures



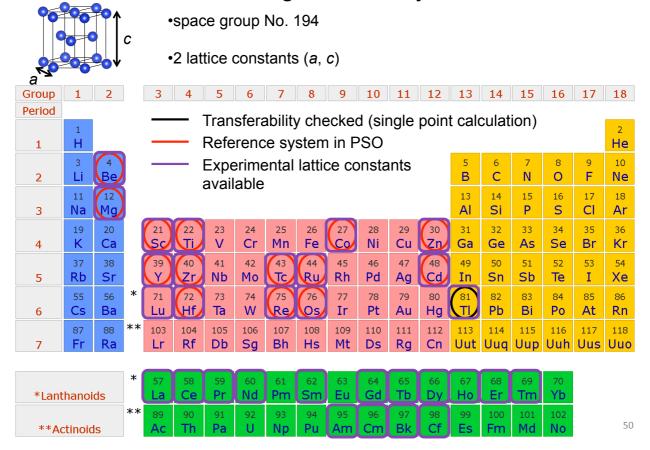
- •space group No. 225
- •1 lattice constant (a)



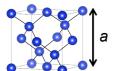
Band structure fitting for SCL crystal structures



Band structure fitting for HCP crystal structures

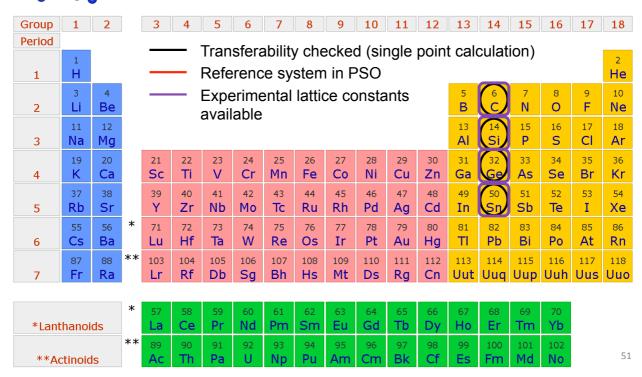


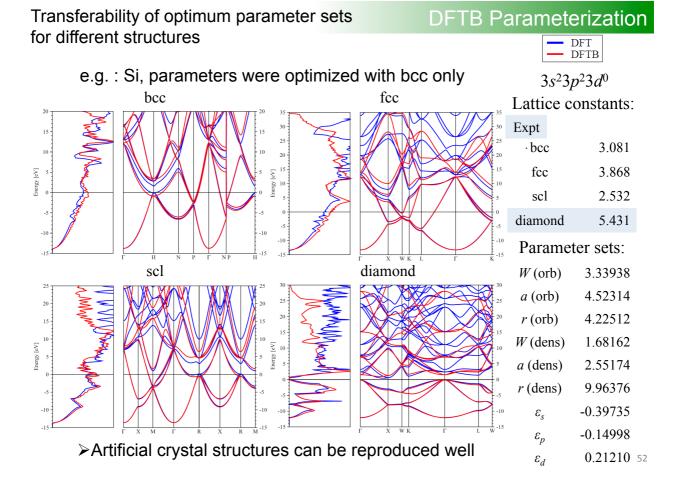
Band structure fitting for Diamond crystal structures



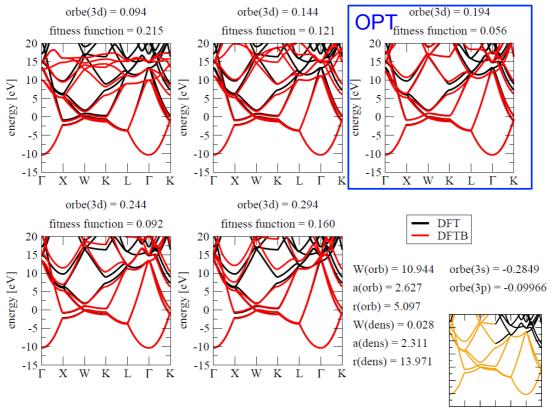
•space group No. 227

•1 lattice constant (a)



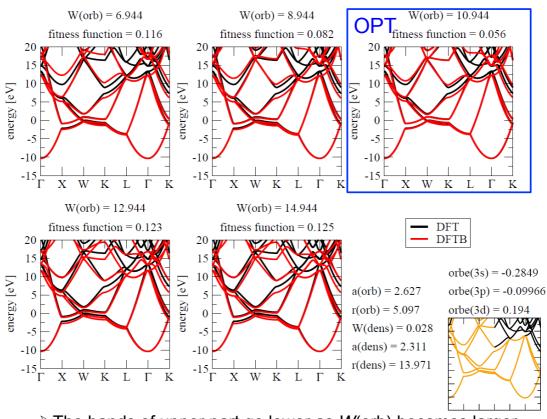


Influence of virtual orbital energy (3d) to Al (fcc) band structure



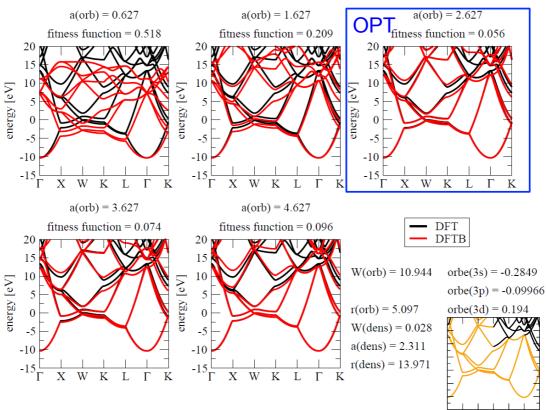
>The bands of upper part are shifted up constantly as orbε(3d) becomes larger

Influence of W(orb) to Al (fcc) band structure



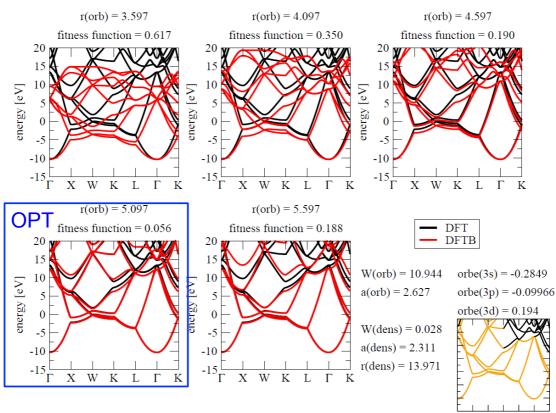
The bands of upper part go lower as W(orb) becomes larger

Influence of a(orb) to Al (fcc) band structure



➤Too small a(orb) gives the worse band structure

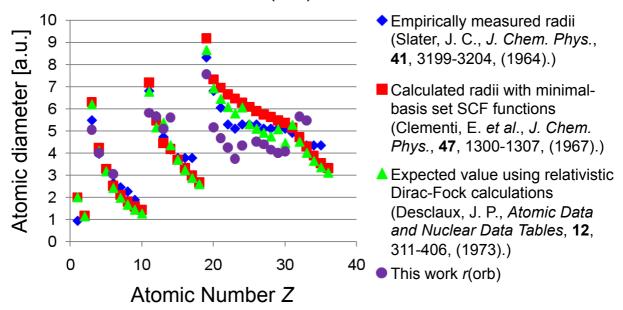
Influence of r(orb) to Al (fcc) band structure



>r(orb) strongly influences DFTB band structure

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Correlation of r(orb) vs. atomic diameter

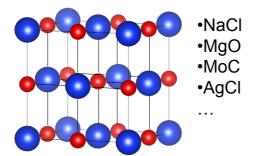


 \triangleright In particular for main group elements, there seems to be a correlation between r(orb) and atomic diameter.

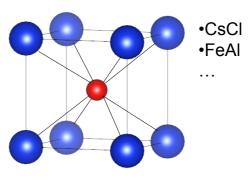
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Straightforward application to binary crystal structures

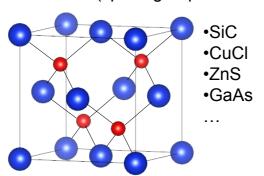
Rocksalt (space group No. 225)



B2 (space group No. 221)



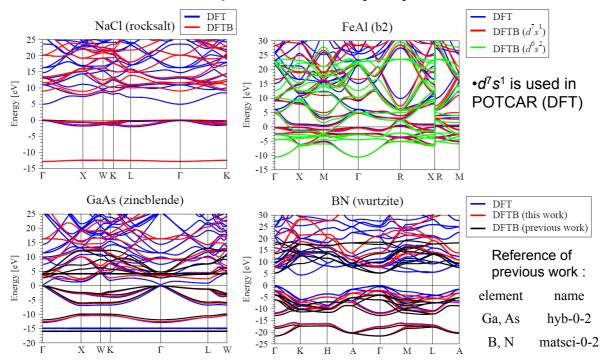
Zincblende (space group No. 216)



Others

- •Wurtzite (BeO, AlO, ZnO, GaN, ...)
- Hexagonal (BN, WC)
- •Rhombohedral (ABCABC stacking sequence, BN)
- > more than 100 pairs tested

Selected examples for binary crystal structures



Further improvement can be performed for specific purpose but this preliminary sets will work as good starting points

Acknowledgements

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- Marcus Elstner
- Jan Knaup
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- Yasuhito Ohta
- Thomas Heine
- Keiji Morokuma
- Marcus Lundberg
- Yoshio Nishimoto
- Some others

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